Instructions: Use whatever format you would like to work on this assignment, but include in the filename the number of this quiz (i.e., 01) and your last name.

## Thermodynamics

1) The article that you read in preparation for this discussion section (available here: https://pubs.rsc.org/en/content/articlelanding/1977/f1/f19777300011) described a so-called Förster cube. What are some of the approximations and/or shortcomings in such a diagram?
2) Pyranine is a hydroxyarene molecule with measured $\mathrm{p} K_{\mathrm{a}}$ and $\mathrm{p} K_{\mathrm{a}}{ }^{*}$ values of $\sim 7.3$ and $\sim 1$, respectively. Assuming that you have an aqueous solution of pyranine at pH 4 at thermal and (electro)chemical equilibrium (i.e. in the dark), and all activity coefficients equal 1 , what is the relative ordering of the concentrations of all four species in the Förster cycle proton-transfer square scheme? Then assume that a constant source of light is introduced and the species are perturbed from equilibrium to a steady-state condition. Assuming that the electronic excited-state lifetimes are the same for the protonated and deprotonated forms of pyranine and are sufficiently long to enable excited-state proton-transfer chemistry to occur, what is the relative ordering of the ability of each species to perform useful non- $p_{\text {system }} \mathrm{d} V_{\text {system }}$ work, i.e. the change in (electro)chemical potential of each species? Explain your answers to each question.
3) While we know that a system at (electro)chemical equilibrium has $\Delta G=0$, how else can we write (electro)chemical equilibrium in terms of mass action rate equations and can we derive the equation for $\Delta G^{\circ}$ using them? Do this for the reaction $2 \mathrm{~A}+\mathrm{B} \rightleftarrows \mathrm{C}$, and assume it to be spontaneous under standard-state conditions, i.e. exergonic/exoergic.
4) Considering that matter has properties of both particles and waves, use the plane wave equation, $\Psi=$ $A e^{i k_{x} x} e^{i k_{y} y} e^{i k_{z} z} e^{-i \omega t}=A e^{i(k \cdot r-\omega t)}$, to derive the operators for energy and momentum used in quantum mechanics.
